



Effects of iron toxicity on germination and initial growth of *Carica papaya* L.

Efeitos da toxicidade por ferro na germinação e crescimento inicial de *Carica papaya* L.

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Iron (Fe) is vital for plant development. Its excessive is harmful, both to forest and cultivated species. This study analyzed the effect of high Fe doses on the germination and initial growth of *Carica papaya* seeds, submitted to concentrations of 0.045, 4, and 8 mM in the form of ferrous sulfate and FeEDTA. Germination percentage, germination speed index, shoot and root growth, fresh and dry mass of the seeds were analyzed. For the initial growth the following variables were analyzed: leaf area, stem and root growth, fresh and dry mass, quantification of chloroplastid pigments, analysis of chlorophyll a fluorescence, element contents and the activity of the enzymes Superoxide Dismutase, Catalase and Peroxidase. Fe stress affected the percentage and speed of germination, root growth, and the accumulation of biomass of *C. papaya*, with FeEDTA being responsible for the most pronounced effects. Fe led to increases in chlorophyll and photosynthetic performance without changes in carotenoid levels. The 8 mM treatments impaired root growth, aerial growth, leaf area expansion, and biomass accumulation. The increase in Fe concentration led to an increment in its deposition in root tissue, with a drop in potassium levels in the 8 mM treatment. There was an increase in Peroxidase activity when submitted to FeEDTA. There was no difference in Superoxide Dismutase and Catalase expression in any treatment. These results show the sensitivity of *C. papaya* species to toxic levels of Fe, which caused damage to metabolism and initial growth.

Keywords: seed, metal, stress

O ferro (Fe) é vital para o desenvolvimento das plantas. Seu excesso é prejudicial para espécies florestais e cultivadas. Este estudo analisou o efeito de altas doses de Fe na germinação e desenvolvimento inicial de *Carica papaya*, submetidas a concentrações de 0,045, 4 e 8 mM na forma de sulfato ferroso e FeEDTA. Porcentagem de germinação, índice de velocidade de germinação, comprimento aéreo e radicular e massa fresca e seca das sementes foram analisadas. Para o desenvolvimento inicial as seguintes variáveis foram analisadas: área foliar, comprimento do caule e raiz, massa fresca e seca, quantificação de pigmentos cloroplastídeos, análise da fluorescência da clorofila a, conteúdo dos elementos, e atividade das enzimas Superóxido Dismutase, Catalase e Peroxidase. O estresse por Fe afetou a porcentagem e velocidade de germinação, crescimento radicular e acúmulo de biomassa de *C. papaya*, sendo o FeEDTA responsável pelos efeitos mais pronunciados. O Fe levou a aumentos no teor de clorofila e no desempenho fotossintético sem alterações nos carotenóides. Os tratamentos com 8 mM afetaram o crescimento radicular e aéreo, expansão da área foliar e acúmulo de biomassa. O aumento na concentração de Fe acarretou uma maior deposição no tecido radicular, com queda nos níveis de potássio no tratamento 8 mM. Houve aumento na atividade da Peroxidase quando submetido ao FeEDTA. Não houve diferença na expressão da Superóxido dismutase e Catalase. Estes resultados mostram a sensibilidade da espécie *C. papaya* ao Fe, causando danos ao metabolismo e desenvolvimento inicial.

Palavras chave: semente, metal, estresse

1. INTRODUCTION

Mining and processing of Fe are today one of the most important economic activities in southeastern Brazil, especially in the states of Minas Gerais and Espírito Santo [1]. However, the extraction of this mineral generates problems such as the production of non-biodegradable waste which affects both the environment near the extraction and beneficiation area and surrounding regions that may be directly or indirectly linked to production [2].

Toxicity caused by increased Fe concentration is characterized by the accumulation of this metal in all plant tissues, causing foliar chlorotic stains, loss in photosynthetic efficiency, enzymatic disturbances, a deficit in biomass accumulation, and decrease in growth and fruiting rate in cultivated species [3, 4].

Losses caused by the increased Fe content in the productivity of these species is a concern that dates back to the middle of the 20th century when excess Fe was verified in Asia in rice plantations [3]. Other works relating Fe stress to the production of cultivated plants include *Nicotiana tabacum* L. [5], *Lactuca sativa* L. and *Eruca sativa* Mill. [6], *Glycine max* L. Merrill [7] and *Coffea* spp L. [8].

The species *Carica papaya* is a tropical fruit tree known by the common name papaia and presents notorious importance in the national economic scenario. The culture occurs in many states of the country, and the city of Linhares, Espírito Santo (Brazil), presents a significant contribution to the national production [9, 10], with 60 thousand tons of fruit sold in the country per year [11].

The city of Linhares is cut by watercourses that historically suffer from an accumulation of heavy metals, especially Fe [12, 13]. In 2015, Linhares main river suffered with the accumulation of Fe due to the disaster of Mariana, in the state of Minas Gerais. Studies confirmed that more than 40 million cubic meters of Fe-rich waste contaminated the environment, limiting the crop productivity of this species [14].

The study of *C. papaya* as an indicator of environmental quality is essential, not only because it has a wide distribution in areas affected by Fe accumulation [15], but also because it is a widely cultivated, commercialized and consumed species in Brazil. This study analyzed the effects of excess Fe on the germination and initial growth of the species *C. papaya*.

2. MATERIALS AND METHODS

Seeds of *C. papaya*, Sunrise solo variety, were stored in the Plant Interactions Laboratory of the Federal University of Espírito Santo in Vitória (ES) in a cold chamber (5 °C). Four treatments (two Fe sources and two concentrations) plus a control treatment, were tested for germination and initial growth. The concentrations used were 0.045 (control), 4 mM, and 8 mM, applied in the form of ferrous sulfate (FeSO₄), and FeEDTA. The experimental design was entirely randomized, with four repetitions of 25 seeds for the germination test and 20 plants per treatment for the analysis of initial growth. The means obtained from the treatments were compared using an analysis of variance (ANOVA) and the Tukey test at 5% significance. All statistical analyses used the Sisvar program version 5.7.

Germination Test

Seeds were disinfected with 2% sodium hypochlorite for 2 minutes, washed and sown in Petri dishes. Then, seeds were lined with two sheets of filter paper moistened with FeSO₄ and FeEDTA solutions corresponding to each treatment [16]. The germination tests were run in a B.O.D. germination chamber, under a constant temperature of 25 °C and 12/12h photoperiod. Germination was monitored daily, considering germinated seeds with radicle protrusion of at least 2 mm [17]. The germination percentage (%G), germination speed index (GSI) [18], root and aerial part growth (cm) [19] and fresh and dry masses (g) were analyzed [20].

Initial growth analysis

Twenty days old plants, growing in unfertilized substrate plus washed sand in a 1:1 ratio, were submitted to the respective treatments with Fe solutions applied via soil. It was analyzed the antioxidant enzyme activity, chloroplastid pigments, chlorophyll a fluorescence, growth measures and elements content. Plants grew under a constant temperature of 25 °C and 12/12h photoperiod and were maintained in Hoagland solution at ionic half strength and pH 5.0 [21].

Antioxidant enzyme activities

The activity of the antioxidant enzymes Superoxide Dismutase (SOD), Peroxidase (POX), and Catalase (CAT) was verified twenty days after the Fe application via soil. For the extraction of the enzymes, 300 mg of root plant material was homogenized with 0.1 M potassium phosphate buffer (pH 6.8), 0.1 mM EDTANa², and 1% polyvinylpyrrolidone (PVPP). The extractions were performed using mortar and pestle with liquid nitrogen. The material was centrifuged at 12000 xg for 15 min at 4 °C. The supernatant was used for the SOD, CAT and POX tests. SOD activity was verified at 25 °C in a 15 W lamp-lit chamber, based on Del Longo et al. (1993) [25], Giannopolitis and Ries (1977) [26], and Beauchamp and Fridovich (1971) [27]. After 6 minutes of exposure, a 560 nm reading was performed; the measured POX activity was according to Kar and Mishra (1976) [28] and Chance and Maehley (1955) [29], and the reaction occurred at room temperature for 2 minutes following the 420 nm reading, and the CAT activity followed the Havir and Mchale (1987) [30] and Anderson, Prasad and Stewart (1995) [31] protocol, and occurred at room temperature with a 240 nm reading for 2 minutes. The design used was entirely randomized with three repetitions with duplicates, according to Peixoto et al. (1999) [32].

Extraction and quantification of chloroplastid pigments

The chlorophyll a, b, and carotenoid contents were verified twenty days after the Fe application via soil, using three replicates with 20 mg of fresh foliar mass each, homogenized in 5 mL of icy (80%) acetone. After that, the material was filtered using a funnel and filter paper, and the filtered liquid was then stored in a 10ml volumetric flask wrapped in aluminum foil and PVC. The pigments were quantified using a spectrophotometer (Genesys 10S UV-Vis, Thermo Fisher Scientific, Waltham, USA) at wavelengths of 470, 645, and 662 nm. according to Lichtenthaler (1987) [22].

Chlorophyll a fluorescence

Simultaneously with the pigment content, the chlorophyll fluorescence was also quantified using a Handy-PEA (Photosynthetic Efficiency Analyser) Hansatech Instruments®, King's Lynn, Norfolk, UK portable fluorometer. The measurements occurred on expanded young leaves of 10 plants per treatment, and previously adapted to 40 minutes of darkness using leaf clips for complete oxidation of the photosynthetic system. The results were presented in an electronic spreadsheet using the PEA Plus v1.11 software. From this analysis, it was calculated the density of photosynthetic reaction centers of the photosystem II (PSII) (RC/ABS), the efficiency with which electrons move ($\delta R0$), the performance of the photosystem I (PSI) oxy reduction reactions ($\delta R0 / (1 - \delta R0)$) and the overall performance of the plant (PIABS + PITOTAL). All are biophysical parameters that quantify the energy flow through the electron transport chain using the JIP test [23].

Growth measures

The growth of the plants was evaluated at 10 and 20 days after the application of Fe solutions, and the parameters used for the analysis were leaf area (cm²), stem and root growth (cm), and fresh and dry mass (g).

Element content with scanning microscopy

Plants were dehydrated in a drying oven at 70 °C for three days. Then, the material was sectioned into leaf, stem, and root, which were then covered with gold for analysis by scanning electron microscopy and then analyzed by Energy Dispersive Spectroscopy (EDS) to detect the elements absorbed by their parts [24].

3. RESULTS AND DISCUSSION

Fe accumulation significantly affected the germination of *C. papaya* species, reducing %G, GSI, root growth, and dry biomass values of germinated seeds. FeEDTA treatment induced the most adverse effects on dry mass accumulation, with a 55% drop in root weight and 42% in airborne weight when compared to control. The same treatment provided an 80% reduction in root growth and a significant delay in GSI, causing seeds to germinate 65% slower than the control. There was no difference in aerial growth between treatments (Table 1).

Table 1: Effects of Fe stress on germination percentage (%G), germination speed index (GSI), root growth (RG), aerial growth (AG), fresh root mass (FRM), fresh aerial mass (FAM), dry root mass (DRM) and dry aerial mass (DAM) of *Carica papaya* seeds.

Treatment	%G	GSI	RG(cm)	AG(cm)	FRM(g)	FAM(g)	DRM(g)	DAM(g)
Control	53a	2.21a	1.86a	1.45ab	0.0078a	0.027a	0.0035a	0.0046a
Fe ⁺² 4 mM	41b	1.73b	1.50a	1.60a	0.0066ab	0.024a	0.0027b	0.0046a
Fe ⁺² 8 mM	40b	1.74b	0.54b	1.16ab	0.0058ab	0.020ab	0.0011c	0.0042a
Fe EDTA 4 mM	23c	0.73c	0.38b	0.82b	0.0043b	0.013b	0.0013c	0.0028b
Fe EDTA 8 mM	25c	0.78c	0.36b	0.80b	0.0032b	0.013b	0.0016c	0.0027b

Different letters in the same column indicate significant differences between treatments. ($p \leq 0.05$, Tukey test).

Although Fe is essential to plant metabolism and participates in crucial processes such as photosynthesis, respiration and nitrogen fixation [33, 34], its excess is harmful as it causes the production of H₂O₂ and other reactive oxygen species (ROS) that will, through oxidative stress, affect almost all cellular functions, including embryonic [35]. The elevation in the oxidation process is highly related to damage to proteins, DNA synthesis, and cell division [36]. Such processes are extremely stimulated after the soaking of seeds and may have been affected by exposure to Fe. The seed is a stage in plant development that presents protection against a wide range of stresses. However, after soaking its sensitivity increases [37] and, in addition to the effects already mentioned, Fe (especially in its chelated version) and other metals can reduce the mitotic activity of the embryonic meristem and present clastogenic effects on chromosomes, as already described in studies involving cultivated species such as corn and sunflower [38, 39]. Fe is also able to directly block the transport of water for the soaking of seeds [40, 41]. This blockage causes direct damage to germination and germination speed, accumulates dangerously in the mitochondria, and thus harms the respiratory metabolism, which may explain the low accumulation of dry matter in FeEDTA treatments when compared to control [42, 43].

Carica papaya plants exposed to high doses of Fe showed a significant increase in POX activity in both concentrations of FeEDTA treatment. There was no alteration on the expression of CAT and SOD enzymes (Table 2).

In response to the accumulation of ROS generated by Fe, the plant's antioxidant mechanisms increase its activity to block possible metabolic damage caused by oxidation [44, 45, 46]. Among these mechanisms, SOD, CAT, and POX enzymes act in the biotransformation of ROS into less harmful compounds to the plant. SOD is the first defense barrier against ROS, while CAT acts almost exclusively on peroxysomes and plant mitochondria. POX have their sites of action preferentially on chloroplasts and cell walls, where they also contribute to the process of cell expansion via oxidation of phenolic compounds in the last step to the formation of lignin [47, 48, 49]. In this work, SOD and CAT activities did not differ between treatments, which does not seem to indicate the presence of significant oxidative stress. However, for POX, it was found that FeEDTA caused an increase in enzyme activity. Increases in its activity have already been

described for other species of cultivars when stressed with Fe, such as *Solanum tuberosum* L. [50], *Zea mays* L. [51] and *Oryza sativa* L. [52, 53]. This fact suggests a certain degree of metabolic disorder in chloroplasts, where there is a higher activity of POXs enzymes and a higher allocation of Fe present in leaves, being, therefore, the organelle that will suffer the most adverse effects when under stress [54].

Table 2: Effects of Fe stress on the activity of the enzymes superoxide dismutase (SOD), peroxidase (POX) and catalase (CAT) in *Carica papaya* plants.

Treatment	SOD (unity. SOD g ⁻¹ DM)	CAT (μmol H ₂ O ₂ min ⁻¹ g ⁻¹ DM)	POX (μmol POX min ⁻¹ g ⁻¹ DM)
Control	0.40a	29.86a	4.76a
Fe ⁺² 4 mM	0.38a	36.19a	9.13a
Fe ⁺² 8 mM	0.35a	33.22a	7.75a
Fe EDTA 4 mM	0.36a	47.42a	19.20b
Fe EDTA 8 mM	0.33a	36.41a	13.50b

Different letters in the same column indicate significant differences between treatments ($p \leq 0.05$, Tukey test).

When analyzing the chloroplastid pigments, no differences were found in carotenoid levels. However, the chlorophyll a concentration increased in all treatments, and this elevation was up to five times when exposed to FeEDTA 8 mM. The chlorophyll b content in the plants varied only when subjected to chelated Fe with a fourteen-fold increase when exposed to 8 mM and compared to control (Table 3).

Table 3: Effects of Fe stress on chlorophyll a, b and carotenoid concentrations of *Carica papaya* plants.

Treatment	Chlorophyll a (mg g ⁻¹ DM)	Chlorophyll b (mg g ⁻¹ DM)	Carotenoids (mg g ⁻¹ DM)
Control	1.50a	0.80a	0.70a
Fe ⁺² 4 mM	3.38b	1.30a	1.07a
Fe ⁺² 8 mM	2.99b	1.01a	0.78a
Fe EDTA 4 mM	4.09c	2.29b	1.02a
Fe EDTA 8 mM	8.1d	11.37c	0.82a

Different letters in the same column indicate significant differences between treatments ($p \leq 0.05$, Tukey test).

Chlorophylls a and b are the main pigments responsible for the capture of light energy that will start the photosynthetic process. It is common to use the concentration of these pigments as an indicator of the effects of several abiotic stresses, including Fe, on plants [55]. Such stresses may inhibit the synthesis of 5-aminolevulinic acid, the precursor molecule of chlorophyll, or increase the activity of the chlorophyllase enzyme, which in turn degrades the chlorophylls present [56]. Also, Fe can block reducing compounds and enzymes that will participate in the pigment biosynthesis route [57]. Nevertheless, in this study, there was a significant increase in chlorophyll levels. Few studies have shown a positive relationship between an increase in heavy metal content and an increase in the amount of chloroplastidic pigments, being restricted basically to green algae [58].

However, Ma et al. (2016) [59], analyzing the effects of different ferrous sulfate concentrations on the growth of *Zea mays* L., found that concentrations up to 500 mg/kg (~4 mM) stimulated the increase in chlorophyll content. Other similar results showed in individuals of the *Jatropha curcas* L. species grown on a high Fe substrate and in young plants of *Anadenanthera colubrina* (Vell.) Brenan and *Moringa oleifera* Lam. in which increased pigment concentrations were observed [60, 61, 62]. This increase may be related to the investment of the plant in adaptation mechanisms, in which the increase in chlorophyll content can promote a faster recovery of photosynthetic activity when the plant returns to non-stress conditions [63]. Pigment content is closely related to photosynthetic quality, and any change in chlorophyll and carotenoid content can affect photosynthesis and plant productivity [57].

The analysis of the effects of high Fe levels on photosynthetic variables using the JIP test showed that there was no significant difference between the density values of active PSII reaction centers, nor the performance of the oxi-reduction reactions of the PSI (Figure 1). On the other hand, for all treatments, when compared to control, it was observed a significant increase in the locomotion efficiency of electrons and the overall performance of the plant (Figure 2).

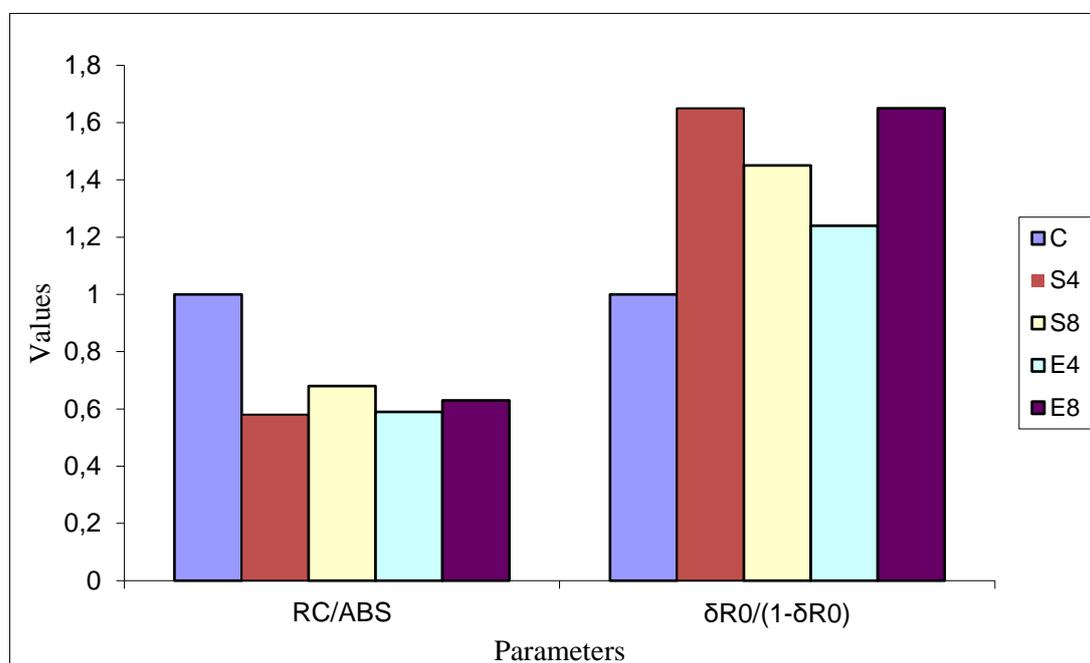


Figure 1: Effects of Fe stress on RC/ABS and $\delta R0/(1-\delta R0)$ of *C. papaya* plants. (C): Control; (S4): Fe^{+2} 4 mM; (S8): Fe^{+2} 8 mM; (E4): FeEDTA 4 mM; (E8): FeEDTA 8 mM. ($p \leq 0.05$, Tukey test).

The increase in electron transport and the performance indexes of *C. papaya* plants were in line with the increase in the content of chloroplastid pigments. The increase in chlorophylls a and b concentration allowed a higher light uptake, which resulted in higher performance of both photosystems, without, however, need to raise the density of the reaction centers. Similar increases in electron transport and photosynthetic performance were found by Santos Junior (2018) [62] in his work with *J. curcas* L. plants equally exposed to high Fe contents in the soil during the initial development phase. Such results may indicate a tolerance of *C. papaya* species to Fe stress in their energy transport to the PSI.

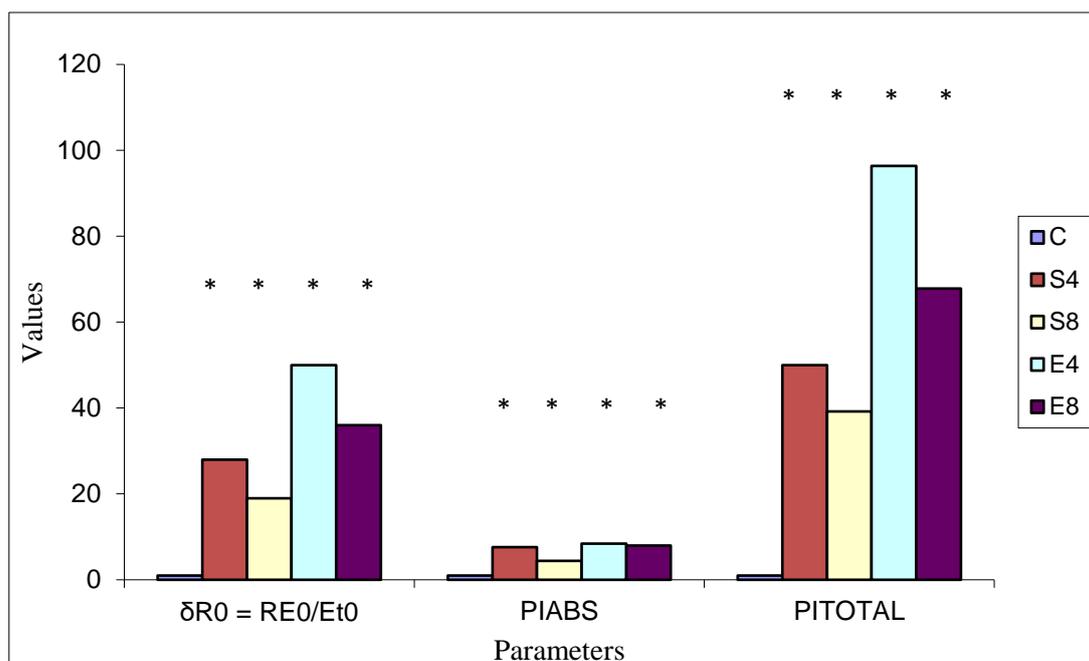


Figure 2: Effects of Fe stress on $\delta R0 = RE0/Et0$, PIABS and PITOTAL of *C. papaya* plants. (C): Control; (S4): Fe^{+2} 4 mM; (S8): Fe^{+2} 8 mM; (E4): FeEDTA 4 mM; (E8): FeEDTA 8 mM. Asterisk indicates significant difference between mean and control. ($p \leq 0.05$, Tukey test).

Exposure to Fe caused a decrease in root size, aerial part, leaf area, and biomass in both time evaluations, i.e., at 10 and 20 days of the experiment. Fe applied via solution to the soil showed inhibitory effects with both sources of the metal. It was observed near 80% drop in leaf area expansion at ten days compared to the control, on the 8 mM concentration of the Fe^{+2} treatment. Similarly, although the same treatment did not provide reductions in root sizes and aerial part at the end of the experiment, it brought significant drops in dry mass accumulation of the aerial part at 10 and 20 days with decreases of 67% and 69% respectively. FeEDTA 8 mM treatment showed even more considerable damage, with a 75% reduction in the dry mass of the aerial part, both at 10 and 20 days. Fe in its chelated form also caused reductions in leaf area measurements and root sizes (for treatment 8 mM at 20 days) and aerial part in both time evaluations, when compared to control. Root dry mass values did not change between treatments (Table 4).

It was verified the deleterious effects of Fe on plants growth from the concentration of 4 mM for the chelated form, and of 8 mM for the ferrous sulfate. Other studies found similar damage caused by Fe excess in plant development. *Eugenia uniflora* L. plants exposed to high Fe concentrations showed a decrease in root growth, plant height, as well as accumulation of dry matter [46]. Neves (2004) [64], also working with *E. uniflora* L., observed a significant decrease in branches and leaves numbers, stem and roots growth, and root volume of plants subjected to stress by ferric citrate. In cultivated species, the same effects on growth were verified in *Oryza sativa* L., *Ipomoea batatas* (L.) Lam., and *Vigna radiata* (L.) R. Wilczek [3, 47, 34]. The toxic effects of Fe may be related to its efficiency in inactivating several key enzymes of plant metabolism, which will culminate in blocking protein and DNA synthesis, as well as causing nutritional disorders [65, 66]. Additionally, Fe can act by reducing the availability of the reducing power of $NADPH_2$ and ATP or causing an imbalance in the enzymes involved in fixing atmospheric CO_2 in the Calvin cycle. In the photosynthetic stage, Fe has already demonstrated the ability to interfere negatively in the process of stomata opening, which will lead to a reduction in the diffusion of CO_2 to mesophiles and a decrease in the activity of the carboxylation phase of photosynthesis [45, 67]. These Fe capacities may explain the non-elevation of the oxi-reduction reactions of the PSI, added to the decrease in the growth rate of the plants, despite the increase in pigment content and photosynthetic performance.

Table 4: Effects of Fe stress on root growth (Root), stem growth (Stem), leaf area (LA), fresh root mass (FRM), fresh aerial mass (FAM), dry root mass (DRM) and dry aerial mass (DAM) of *C. papaya* plants at 10 and 20 days after the beginning of treatments.

Treatment	Root (cm)		Stem (cm)		LA (cm ²)		FRM (g)		DRM (g)		FAM (g)		DAM (g)	
	10d.	20d.	10d.	20d.	10d.	20d.	10d.	20d.	10d.	20d.	10d.	20d.	10d.	20d.
Control	2.88a	3.34a	4.62a	4.84a	2.70a	3.17a	0.047a	0.068a	0.0015a	0.0029a	0.147a	0.150a	0.0110a	0.0160a
Fe ⁺² 4 mM	2.36ab	3.26a	3.58b	4.58a	0.85b	2.95ab	0.028ab	0.064a	0.0016a	0.0021a	0.054b	0.130a	0.0041b	0.0130a
Fe ⁺² 8 mM	0.86b	2.38ab	2.90bc	4.52a	0.44c	1.95bc	0.011b	0.036b	0.0014a	0.0023a	0.038b	0.070ab	0.0037b	0.0050b
Fe EDTA 4 mM	2.08ab	2.32ab	2.96bc	2.70b	0.64bc	0.96c	0.026b	0.030b	0.0014a	0.0026a	0.046b	0.044b	0.0030b	0.0054b
Fe EDTA 8 mM	2.26ab	1.96b	2.52c	2.68b	0.55c	1.03c	0.018b	0.031b	0.0010a	0.0023a	0.040b	0.034b	0.0026b	0.0041b

Different letters in the same column indicate significant differences between treatments. ($p \leq 0.05$, Tukey test).

The content of constituent elements analysis after 20 days, showed little Fe translocation to the aerial part. It was verified the highest metal contents in the roots of plants submitted to treatments of 8 mM in sulfate form and EDTA, with 4.12% and 3.45% of dry weight, respectively, which corresponds to an increase of 250% in root Fe levels when compared to treatments in the 4 mM concentration (Supplementary file). The high Fe levels present in the root region may be part of a strategy observed in some plants, which consists in the allocation of Fe in the vacuole of non-photosynthesizing tissues or with low production, aiming at protecting the primary metabolism [68, 69]. This process is concomitant with other mechanisms that, in an attempt to block the entrance of the metal, modify the pH of the region close to the root, favoring the oxidation of Fe^{+2} in non-absorbable Fe^{+3} , causing the formation of Fe plates [70].

In the same root region, it was possible to notice the potassium (K) levels at doses of 8 mM dropped 70% in ferrous sulfate and 50% in chelated Fe, compared to the control (Supplementary file F and J). Stress, whether biotic or abiotic, is a recognized agent disturbing the metabolism of K [71, 72]. Also, it is known that high Fe concentrations affect the balance of various ions, especially K [73]. Some studies show the proportional relationship between high Fe levels and significant ion loss [74, 75, 76, 77]. Excess Fe leads to the formation and accumulation of nitric oxide (NO) in root cells, and NO production responds significantly to increased exposure to Fe [78, 79]. NO can stimulate the activation of non-selective cationic channels (NSCCs), which are primarily responsible for K efflux [73, 80]. K is an essential osmotic cell activator, acting crucially in opening and closing the stomata and in protein synthesis. The drop in its content in 8 mM treatments may be one of the possible causes of the damage that compromised the growth of *C. papaya*.

4. CONCLUSIONS

Fe, both in its reduced and chelated form, strongly inhibits germination, germination speed, root growth, and the accumulation of biomass of *C. papaya* seeds. FeEDTA can stimulate the activity of the enzyme POX, but without indications of oxidative stress in *C. papaya* plants. High Fe content causes elevation of chlorophyll a, b, and photosynthetic performance of *C. papaya*, but affected initial growth with a decrease in the size of the radicle, aerial part, leaf area, accumulation of biomass, and levels of K. Thus, the productivity of this species is severely limited in areas affected by Fe accumulation.

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